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Periodic Classification of Elements CHAPTER5 In Class IX we have learnt that matter around us is present in the form of elements, compounds and mixtures and the elements contain atoms of only one type. Do you know how many elements are known till date?

CHAPTER5 Periodic Classification of Elements

Modern Periodic Table → Atomic number of an element is a more fundamental property than its atomic mass. → According to the Modern Periodic law : The properties of elements are a periodic function of their atomic number. → All the anomalies of Mendeleev’s classification disappear.

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Chapter 5 - Periodic Classification of Elements Exercise 85 Solution 1 K is in group 1. Therefore, the oxide will be K 2 O.

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Video Timeline: 0:00:00 Starts 0:08:17 Introduction 0:11:34 Classification of Periodic Table- 1) Dobereiner's Triads 0:19:40 2) Newland's Law of Octaves 0:26...

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