

## Access Free Chapter 14 Acids And Bases Worksheet

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## **Chapter 14 Acids And Bases**

an acid that is composed of H, O, and a third element. The H & O are covalently bonded and are called the hydroxyl group. Property of bases. taste bitter, feel slippery, change color of indicators, react with acids to produce a salt and water neutralization rxn, are electrolytes. Arrhenius or traditional base.

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(14.1) The nature of acids and bases

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(14.2) Acid strength (14.3) The pH scale (14.4) Calculating the pH of strong acid solutions (14.5) Calculating the pH of weak acid solutions (14.6) Bases (14.7) Polyprotic acids

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Chapter 14 - Acids and Bases . 14.1 The Nature of Acids and Bases . A. Arrhenius Model 1. Acids produce hydrogen ions in aqueous solutions 2. Bases produce hydroxide ions in aqueous solutions B. Bronsted-Lowry Model 1. Acids are proton donors 2. Bases are proton acceptors 3. H<sub>3</sub>O<sup>+</sup> is called the hydronium ion C. Conjugate Acid- Base Pairs 1.

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## **Chapter 14 - Acids and Bases - ScienceGeek.net**

Chapter 14 - Acids and Bases.

Description. Chapter 14 Review. Total Cards. 36. Subject. Chemistry. Level. 11th Grade. ... General Properties of Acids: Definition. 1) sour taste, corrosive 2) characteristically changes acid-base indicator 3) react with active metals to form hydrogen gas 4) react with bases to form salt and water 5) conduct ...

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ANS: e)  $\text{ClO}_4^-$ ,  $\text{F}^-$ ,  $\text{OAc}^-$ ,  $\text{CN}^-$  PAGE: 14.2  
11. The conjugate base of a weak acid is  
a) a strong base b) a weak base c) a strong acid d) a weak acid e) none of these  
ANS: b) a weak base PAGE: 14.2

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12. Which of the following is the equilibrium constant expression for the dissociation of the weak acid HOCl? a)  $K = [H^+][OCl^-] [HOCl]$

### **CHAPTER 14 Acids and Bases**

Modern Chemistry Chapter 14 Vocab  
Acids and Bases. binary acid. oxyacid.  
Arrhenius acid. Arrhenius base. an acid that contains only two different elements: hydrogen an.... an acid that is a compound of hydrogen, oxygen, and a third el.... a chemical compound that increases the concentration of hydrog....

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CHAPTER 14 REVIEW Acids and Bases  
SECTION 1 SHORT ANSWER Answer the

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following questions in the space provided. 1. Name the following compounds as acids: sulfuric acid a.  $H_2SO_4$  sulfurous acid b.  $H_2SO_3$  hydrosulfuric acid c.  $H_2S$  perchloric acid d.  $HClO_4$  hydrocyanic acid e. hydrogen cyanide 2. H

## 14 Acids and Bases - Password

acid, and water acts as the base. This is the acid ionization reaction.  $HA(aq) + H_2O(l) \rightleftharpoons A^-(aq) + H_3O^+(aq)$  6) When a base is added to water, there is a Bronsted-Lowry acid base reaction where the base behave as a base and water acts as the acid. This is the base ionization reaction. Be able to identify or provide the conjugate acids and ...

## Chapter 14: Acids and Bases

CHAPTER 14 ACIDS AND BASES 501 24. A Lewis acid must have an empty orbital to accept an electron pair, and a Lewis base must have an unshared pair of electrons. 25.  $NH_3 + NH_3 \rightleftharpoons NH_2^- + NH_4^+$  Acid Base Conjugate Conjugate

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Base Acid

## **CHAPTER 14 ACIDS AND BASES - HCC Learning Web**

Major topics: Arrhenius vs. Bronsted-Lowry definition of acids and bases, conjugate acid/base, acid dissociation constant ( $K_a$ ), & strong vs weak acids

## **Chapter 14 (Acids and Bases) - Part 1 - YouTube**

Chapter Fourteen: Acids and Bases  
Chapter interactives: Acids/bases and indicators  
Chemistry battleship  
Chemistry jokes  
pH simulation  
Chapter Homework: Section 1: Chapter review 1 thru 7. Section 2: Chapter review 12, 13. Section 3: Chapter review 19 thru 25.

## **Chapter Fourteen [Acids and Bases]**

Chapter 14: Acids and Bases  
1. Acids and Bases.  
1. Which of the following is the equilibrium constant expression for the dissociation of the weak acid  $\text{HOCl}$ ?  
e) none of these.  
2. In which of the following reactions does the  $\text{H}_2\text{PO}_4^-$  ion

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act as an acid? d)The ion cannot act as an acid.

## **Chapter 14: Acids and Bases 1 - Docest**

Chapter 14 - Acids and Bases Pablo Gonzalez. Loading... Unsubscribe from Pablo Gonzalez? ... Chapter 16 Acid-Base Equilibria - Duration: 1:06:16. Michael Farabaugh 22,088 views.

## **Chapter 14 - Acids and Bases**

Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In this chapter you will learn The definitions and identity of substances that are called acids  
Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In...

## **Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In ...**

Acids and Bases: Chapter 14 & 15 . HW:  
•Read Ch 14: •Fill in as much of the acid base table as you can, as you read . Acid base conductivity and reactivity  
Conducvity, Reac'vity,,



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Hydrochloric\*acid\* high high  
Acidic\*acid\* \*\* low\* medium  
Dis4lled\*water\* none\* none\*  
Ammoniumhydroxide med\* none\*  
Sodiumhydroxide\* high none\* \*

## **Acids and Bases: Chapter 14 & 15**

Friday, January 19 - Day A Agenda: 1)  
Test - Chapter 12 and 13 HW: 1) Watch  
Videos and take notes on intro to acids  
and bases , and pH scale 2) Read 14.1-  
14.4

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