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A molecule is an electrically neutral group of two or more atoms held together by chemical bonds. Molecules are distinguished from ions by their lack of electrical charge.. In quantum physics, organic chemistry, and biochemistry, the distinction from ions is dropped and molecule is often used when referring to polyatomic ions.. In the kinetic theory of gases, the term molecule is often used ...

Molecule - Wikipedia

Robert J. Ouellette, J. David Rawn, in Principles of Organic Chemistry, 2015
Electronegativity. Electronegativity is a measure of the attraction of an atom for bonding electrons in molecules compared to that of other atoms. The electronegativity values devised by Linus Pauling, an American chemist, are

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dimensionless quantities that range from slightly less than one for the alkali metals to a ...

Electronegativity - an overview | ScienceDirect Topics

A phrase that applies to covalent bonding and not other kinds of bonds is ... A. electron-sharing. ... Pearson Chemistry Matta, Staley, Waterman, Wilbraham. 3,682 explanations. Chemistry 9th Edition Zumdahl. ... MasteringBiology Chapter 3: The Molecules of Cells. 138 terms. kevinrhurst. Sets with similar terms.

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The octet rule is a chemical rule of thumb that reflects the theory that main-group elements tend to bond in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas. The rule is especially applicable to carbon, nitrogen, oxygen, and the

